

Pumped medium														Special materials / remark		
		GG-25	GGG-40/GGG-40.3	GS-C 25	ERN	Ni-Resist D2	NORTHARD® NH 153	NORILLOY® NL 25 2	1.4008	1.4308	1.4408	1.4500	NORICITD® 9.4306	NORIDUR® 9.4460	NORICLOR® NC 24 6	G-CUAl 10 Ni
Acetic acid Concentration	CH <sub>3</sub> COOH	-	-	-	-	-	-	-	O	+	+	+	-	+	+	+
	10% ,cold	-	-	-	-	-	-	-	O	+	+	+	-	+	+	+
	100 °C	-	-	-	-	-	-	-	O	+	+	+	-	+	+	+
	40% ,cold	-	-	-	-	-	-	-	O	+	+	+	-	+	+	+
	100 °C	-	-	-	-	-	-	-	-	+	+	+	-	+	+	+
	80% ,cold	-	-	-	-	-	-	-	O	+	+	+	-	+	+	+
Acetic aldehyde	CH <sub>3</sub> -CHO	-	-	-	-	-	-	-	+	+	+	+	-	+	+	+
Acetic anhydride	(CH <sub>3</sub> -CO) <sub>2</sub> O	-	-	-	-	-	-	-	O	O	+	O	-	O	O	-
Acetic pentyester	CH <sub>3</sub> CaOC <sub>5</sub> H <sub>11</sub>	-	-	-	-	+	-	-	+	+	+	+	-	+	+	+
Acetone	CH <sub>3</sub> COC <sub>3</sub> H <sub>7</sub>	+	+	+	+	+	-	-	+	+	+	+	-	+	+	inflammable
Acrylic acid	CH <sub>2</sub> =CHCOOH	-	-	-	-	-	-	-	+	+	+	+	-	+	-	Polymerized, light
Acrylonitrile	CH <sub>2</sub> =CHCN	+	+	+	+	+	-	-	+	+	+	+	-	+	+	-
Alkazid solution,	Cold	-	-	-	-	-	-	-	+	+	+	+	-	+	O	
	hot	-	-	-	-	-	-	-	O	+	O	-	-	+	-	
Aluminum acetate	Al(CH <sub>3</sub> COO) <sub>3</sub>	-	-	-	O	-	-	-	+	+	+	O	-	+	+	
Aluminum chloride	AlCl <sub>3</sub>	-	-	-	O	-	-	-	-	O	-	-	-	+	O	
Aluminum hydroxide suspension	-	-	-	-	-	-	-	+ O	-	-	-	-	-	-	to 80 °C	
Aluminum sulphate	Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	-	-	-	O	-	-	-	O	O	+	+	-	+	O	
Ammonia, cold	NH <sub>3</sub>	-	+	+	+	-	-	-	+	+	+	+	-	+	-	
Ammonia water, boiling	NH <sub>3</sub> ·OH	-	-	-	O	-	-	-	O	O	+	+	-	-	+	
Ammonium carbonate	(NH <sub>4</sub> ) <sub>2</sub> CO <sub>3</sub>	-	-	-	O	-	-	-	O	+	+	+	-	+	-	
Ammonium chloride	NH <sub>4</sub> Cl	-	-	-	+	-	-	-	-	+	+	+	-	+	-	
Ammonium hydroxide	NH <sub>4</sub> Cl	-	-	-	-	-	-	-	-	-	-	-	-	-	See ammonia	
Ammonium nitrate, cold	NH <sub>4</sub> NO <sub>3</sub>	-	-	-	O	-	-	-	+	+	+	O	-	+	-	
Ammonium oxalate	(COONH <sub>4</sub> ) <sub>2</sub>	-	-	-	O	-	-	-	+	+	+	+	-	+	-	
Ammonium sulfate	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	-	-	-	O	-	-	-	O	+	+	+	-	+	-	
Ammonium sulfite	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>3</sub>	-	-	-	O	-	-	-	O	+	+	+	-	+	-	
Ammonium thiocyanate, cold	NH <sub>4</sub> SCN	-	-	-	O	-	-	-	O	+	+	+	-	+	-	
Amyl acetate	CH <sub>3</sub> COOC <sub>5</sub> H <sub>11</sub>	-	-	-	-	-	-	-	-	-	-	-	-	-	See acetic pentyester, inflammable	
Amyl alcohol	C <sub>5</sub> H <sub>11</sub> OH	-	-	-	-	-	-	-	-	-	-	-	-	-	See pentanol, inflammable	
Aniline(=aminobenzene)	C <sub>6</sub> H <sub>5</sub> NH <sub>2</sub>	O	O	O	-	+	-	-	+	+	+	+	+	O	Temperature must be known, inflammable	
Asphalt	-	+	+	+	+	+	+	+	+	+	+	+	+	+	+	
Barium chloride	BaCl <sub>2</sub>	-	-	-	O	-	-	-	O	O	-	-	O	-	O	
Bauxite suspension	-	-	-	-	-	-	-	+ O	-	-	-	-	-	-	To 80 °C	
Beer	-	O	O	O	O	-	-	-	+	+	+	+	-	+	+	
Beer mash	-	+	+	+	-	-	-	-	+	+	+	+	-	+	+	for increased degrees of purity use 1.4408/9.4460
Beer wort	-	+	+	+	-	-	-	-	+	+	+	+	-	+	+	
Benzenesulfonic acid	C <sub>6</sub> H <sub>5</sub> SO <sub>3</sub> H	-	-	-	-	-	-	-	O	+	+	+	-	+	O	
Benzoic acid, aqueous	C <sub>6</sub> H <sub>5</sub> COOH	-	-	-	-	-	-	-	O	+	+	+	-	+	-	
Benzol	C <sub>6</sub> H <sub>6</sub>	+	+	+	+	+	-	-	+	+	+	+	-	+	+	inflammable
Benzyl acetate (=benzoic ether)	CH <sub>3</sub> -COO-CH <sub>2</sub> -C <sub>6</sub> H <sub>5</sub>	O	O	O	O	-	-	-	+	+	+	+	-	+	O	inflammable
Black liquor	-	-	-	-	-	-	-	-	-	-	-	-	-	-	Precise analysis is necessary	
Bleaching liquor	-	-	-	-	-	-	-	-	-	-	-	-	-	-	see sodium hypochlorite	
Bonderizing solution	-	-	-	-	-	-	-	-	O	+	+	+	-	+	-	
Boric acid, aqueous	H <sub>3</sub> BO <sub>3</sub>	-	-	-	-	-	-	-	O	0	+	+	-	+	O	
Brine (anolyte with >30 mg Cl/l)	-	-	-	-	-	-	-	-	-	-	-	-	O	O	-	titanium, hard rubber(contact ASK)
Butadiene, liquid	CH <sub>2</sub> =CH-CH=CH <sub>2</sub>	O	O	+	+	+	-	-	+	+	+	+	-	+	+	inflammable
Butane (liquefied gas)	C <sub>4</sub> H <sub>10</sub>	O	O	+	+	+	-	-	+	+	+	+	-	+	+	inflammable
Butanol(Butanol 1)	C <sub>4</sub> H <sub>9</sub> CH <sub>2</sub> OH	O	O	O	+	-	-	-	+	+	+	+	-	+	+	inflammable
Butyl acetate	CH <sub>3</sub> COOC <sub>5</sub> H <sub>11</sub>	O	O	O	+	-	-	-	+	+	+	+	-	+	O	inflammable
Butyric acid	CH <sub>3</sub> -CH <sub>2</sub> -CH <sub>2</sub> -COOH	-	-	-	-	-	-	-	O	O	+	+	-	+	O	
Calcium bisulfite base	Ca(HSO <sub>3</sub> ) <sub>2</sub>	-	-	-	-	-	-	-	O	+	+	+	-	+	O	
Calcium chloride	CaCl <sub>2</sub>	-	-	-	-	-	-	-	O	-	-	O	-	O	O	hard rubber, titanium, Hastelloy
Calcium hydroxide	Ca(OH) <sub>2</sub>	O	O	O	-	+	-	-	O	+	+	+	-	-	-	
Calcium hypochloride	Ca(ClO) <sub>2</sub>	-	-	-	-	-	-	-	-	O	-	O	-	O	O	hard rubber, titanium, Hastelloy C
Calcium nitrate	Ca(NO <sub>3</sub> ) <sub>2</sub>	-	-	-	-	+	-	-	+	+	+	+	-	+	-	
Calcium sulfate suspension	CaSO <sub>4</sub> + solid matter	-	-	-	O	-	O	O	-	-	-	-	O	O	-	
Carbon dioxide, aqueous solution	H <sub>2</sub> CO <sub>3</sub>	-	-	-	-	-	-	-	+	+	+	+	-	+	O	
Carbon disulfide	CS <sub>2</sub>	+	+	+	+	+	-	-	+	+	+	+	-	+	-	inflammable
Carbon tetrachloride	CCl <sub>4</sub>	+	+	+	+	+	+	-	+	+	+	+	-	+	+	anhydrous and acid-free
Caustic lime	Ca(OH) <sub>2</sub>	-	-	-	-	-	-	-	-	-	-	-	-	-	-	see calcium hydroxide
Caustic potash	KOH	-	-	-	-	-	-	-	-	-	-	-	-	-	-	see potassium hydroxide
Caustic soda	NaOH	-	-	-	-	-	-	-	-	-	-	-	-	-	-	see sodium hydroxide
Chlorehane (ADIP)	C <sub>2</sub> H <sub>5</sub> Cl	+	+	+	+	+	-	-	+	+	+	+	-	+	+	only, if anhydrous
Chlorine, dry	Cl <sub>2</sub>	O	+	+	+	+	-	-	+	+	+	+	-	-	-	≤ 30 °C
	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	Hard rubber (≤ 80 °C), titanium, Hastelloy C (≤ 50 °C)
Chlorobenzene	C <sub>6</sub> H <sub>5</sub> Cl	O	O	O	O	-	O	O	O	+	-	O	-	-	-	inflammable
Chloroform	CHCl <sub>3</sub>	O	+	+	+	-	-	-	+	+	+	+	-	+	+	anhydrous and acid-free
Chlorosulfonic acid, aqueous solution	SO <sub>3</sub> (OH)Cl	-	-	-	-	-	-	-	-	-	-	-	-	-	-	Hastelloy B
Chromid acid, pure Concentration	CrO <sub>3</sub>	-	-	-	-	-	-	-	+	+	+	+	-	+	-	heed isocorrosion curves!
10% ,cold boiled	-	-	-	-	-	-	-	-	O	-	-	+	+	-	-	
50% ,cold boiled	-	-	-	-	-	-	-	-	-	O	O	+	O	O	-	
+ H <sub>2</sub> SO <sub>4</sub> cold boiled	-	-	-	-	-	-	-	-	-	-	O	+	+	-	-	Si-cast iron
Citric acid, aqueous solution	(CH <sub>3</sub> COOH).C(OH).COOH	-	-	-	-	-	-	-	O	O	+	+	O	+	O	
Cooking lye (calcium bisulfate solution)	Ca(HSO <sub>3</sub> ) <sub>2</sub>	-	-	-	-	-	-	-	-	-	+	+	-	+	+	
Cooper sulfate	CuSO <sub>4</sub>	-	-	-	-	-	-	-	O	O	+	+	-	+	O	
Cooper sulfate + 4% H <sub>2</sub> SO <sub>4</sub>	-	-	-	-	-	-	-	-	O	O	+	+	-	+	-	

Special materials / remark																
Pumped medium		GG-25	GGG-40; GGG-40.3	GS-C-25	ERN	Ni-Resist D2	NORTHARD® NH 15.3	NORLOY® NL 25.2	1.4008	1.4308	1.4408	1.4500	NORCICID® 9.4306	NORDUR® 9.4460	NORCLOR® NC 24.6	G-CuAl 10 Ni
Cresol	C <sub>6</sub> H <sub>5</sub> (CH <sub>3</sub> )OH	O	O	O	O	-	-	+	+	+	+	-	+	+	+	O
Cyclohexanol	(CH <sub>3</sub> ) <sub>2</sub> COH	O	O	+	-	-	-	+	+	+	+	-	+	+	+	inflammable
Cyclohexanone	(CH <sub>3</sub> ) <sub>2</sub> CO	O	O	-	-	-	-	+	+	+	+	-	+	+	+	inflammable
Dichlor ethylene, anhydrous	CHCl=CHCl	+	+	-	-	-	-	+	+	+	+	-	+	+	+	-
Dichloro benzene	C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub>	O	O	O	O	-	-	O	O	O	+	-	+	+	O	inflammable
Diethanolamine (DEA)	(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> NCH <sub>2</sub> CH <sub>2</sub> OH	+	+	-	-	-	-	+	+	+	+	-	+	+	-	for reasons of purity: 1.4408/9.4460
Dimethyl formamide	HCON(CH <sub>3</sub> ) <sub>2</sub>	O	O	-	-	-	-	+	+	+	+	-	+	+	-	inflammable
Dimethyl terephthalate(DMT)	C <sub>6</sub> H <sub>4</sub> (COOCH <sub>3</sub> ) <sub>2</sub>	-	-	-	-	-	-	O	+	+	+	-	+	+	O	molten
Dinitrobenzene, aqueous	C <sub>6</sub> H <sub>4</sub> (NO <sub>2</sub> ) <sub>2</sub>	-	-	-	-	-	-	O	+	+	+	-	+	+	-	-
Dye bath (basic or acid)	O	O	O	O	O	O	O	O	O	O	O	O	O	O	O	Contact ASK for combination
Ether (diethyl ether)	(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> O	+	+	+	-	-	-	+	+	+	+	-	+	+	+	highly inflammable
Ethyl acetate	CH <sub>3</sub> COOC <sub>2</sub> H <sub>5</sub>	+	+	+	-	-	-	+	+	+	+	-	+	+	+	-
Ethyl alcohol (ethanol)	C <sub>2</sub> H <sub>5</sub> OH	O	+	+	-	-	-	+	+	+	+	-	+	+	+	inflammable
Ethylamine	C <sub>2</sub> H <sub>5</sub> NH <sub>2</sub>	-	-	-	O	-	-	+	+	+	+	-	+	+	-	-
Ethylene, liquid	CH <sub>2</sub> =CH <sub>2</sub>	-	-	-	-	-	-	-	-	-	-	-	-	-	-	unalloyed cold-tough cast steel, must be cold-tough at -140 °C
Ethylene carbonate	(CH <sub>2</sub> ) <sub>2</sub> CO <sub>3</sub>	+	+	+	-	-	-	+	+	+	+	-	+	+	+	-
Ethylene glycol	CH <sub>2</sub> OH-CH <sub>2</sub> OH	+	+	+	+	-	-	+	+	+	+	-	+	+	+	-
Ethylene oxide	(CH <sub>2</sub> ) <sub>2</sub> O	O	+	+	+	+	+	+	+	+	+	-	+	+	+	inflammable
Fatty acid	C <sub>n</sub> H <sub>2n+1</sub> (COOH)	-	-	-	-	-	-	-	O	O	+	-	+	+	-	-
Fatty alcohol and fats	O	O	+	+	-	-	-	+	+	+	+	-	+	+	+	Inflammable
Formaldehyde 40%,aqueous solution	CH <sub>2</sub> O	-	-	-	-	-	-	O	+	-	-	-	-	-	-	-
Formic acid	HCOOH													Heed isocorrosion curves!		
Concentration	10%, Cold 90°C	-	-	-	-	-	-	-	+	+	+	+	+	+	+	>70°C Hastelloy C
	60%, Cold 60°C	-	-	-	-	-	-	-	+	+	+	+	+	+	+	
	90%, Cold 60°C	-	-	-	-	-	-	-	O	+	+	+	+	+	+	
	-	-	-	-	-	-	-	-	O	O	+	+	+	+	+	
	-	-	-	-	-	-	-	-	O	O	+	+	+	+	+	
Freon, frigen 12	F12:CF <sub>2</sub> Cl <sub>2</sub>	-	O	O	-	-	-	-	+	+	+	-	+	+	-	-
Fruit juices		-	-	-	-	-	-	-	+	+	+	-	+	+	-	-
Fuel oil	O	O	+	O	+	O	O	O	+	+	+	-	+	+	O	Inflammable
Furfural	(CH <sub>3</sub> ) <sub>2</sub> O.C.CH <sub>3</sub> O	O	O	O	O	O	O	O	+	+	+	-	+	+	O	Inflammable
Furfuryl alcohol	(CH <sub>3</sub> ) <sub>2</sub> O.C.CH <sub>2</sub> OH	O	O	O	O	O	O	O	+	+	+	-	+	+	O	Inflammable
Glucose	C <sub>6</sub> H <sub>12</sub> O <sub>6</sub>	+	+	+	+	+	+	+	+	+	+	-	+	+	+	-
Glycerin	CH <sub>2</sub> OH-CHOH-CH <sub>2</sub> OH	O	O	+	+	+	+	+	+	+	+	-	+	+	+	-
Glycol (ethylene glycol)	(CH <sub>2</sub> OH) <sub>2</sub>	O	O	O	+	+	+	+	+	+	+	-	+	+	+	-
Green base																Precise analysis is necessary
Heat transfer fluid oils		O	O	+	O	+	O	+	+	+	+	+	+	+	O	-
Hexane	C <sub>6</sub> H <sub>14</sub>	O	O	+	+	+	+	+	+	+	+	+	+	+	O	Inflammable
Hydrobromic acid	HBr	-	-	-	-	-	-	-	-	-	-	-	-	-	-	Hastelloy B
Hydrochloric acid, aqueous solution	HCl															See hydrochloric acid
Hydrochlorid acid	HCl	-	-	-	-	-	-	-	-	-	-	-	-	-	-	Hard rubber <80 °C
Hydrocyanic acid	HCN	O	O	O	O	O	O	O	O	O	O	O	O	O	O	-
Hydrofluoric acid,aqueous solution	HF	-	-	-	-	-	-	-	-	-	-	-	-	-	-	Hard rubber (<40 °C)
Hydrogen peroxide	H <sub>2</sub> O <sub>2</sub>	-	-	-	-	-	-	-	+	O	O	+	O	O	-	-
Iron (III) chloride	FeCl <sub>3</sub>	-	-	-	-	-	-	-	-	-	-	-	-	-	-	Hard rubber (<80 °C)
Iron (II) sulphate	Fe <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	-	-	-	-	-	-	-	O	+	+	-	+	+	O	-
Isobutyl alcohol	(CH <sub>3</sub> ) <sub>2</sub> CH.CH <sub>2</sub> OH	O	O	+	+	+	+	+	+	+	+	-	+	+	+	Inflammable
Isopropyl alcohol	(CH <sub>3</sub> ) <sub>2</sub> CHOH	O	O	+	+	+	+	+	+	+	+	-	+	+	+	Inflammable
Kerosene		O	O	+	+	+	+	+	+	+	+	-	+	+	O	Inflammable
Lactic acid	CH <sub>3</sub> -CHOH.COOH	-	-	-	-	-	-	O	O	+	-	-	+	+	O	-
Lactic acid (nitrating acid)	HNO <sub>3</sub> +H <sub>2</sub> SO <sub>4</sub>	-	-	-	-	-	-	O	O	O	O	O	O	O	O	Analysis is necessary
latex		+	+	+	+	+	+	+	+	+	+	-	+	+	+	-
Laurin lactam		-	-	-	-	-	-	O	O	+	-	-	+	+	O	-
Linseed oil		O	O	+	O	+	O	O	O	+	+	-	+	+	O	Inflammable
Magnesium chloride	MgCl <sub>2</sub>	-	-	-	O	-	-	-	O	O	O	O	O	O	O	Hard rubber (<80 °C)
Magnesium sulfate	Mg <sub>2</sub> SO <sub>4</sub>	-	-	-	+	-	-	-	+	-	-	+	+	+	O	-
Maleic acid	(HOOC) <sub>2</sub> .(CH) <sub>2</sub>	-	-	-	-	-	-	O	+	-	-	+	+	-	-	-
Manganese chloride	MnCl <sub>2</sub>	-	-	-	O	-	-	O	O	-	O	O	O	O	O	Hard rubber (<80 °C)
Mercury	Hg	+	+	+	+	+	+	+	-	+	-	+	+	+	-	-
Methanol	CH <sub>3</sub> OH	O	O	+	+	+	+	+	-	+	-	+	+	+	+	Inflammable
Methyl acetate	CH <sub>3</sub> COOCH <sub>3</sub>	-	-	-	-	-	-	O	+	-	-	+	+	O	-	Inflammable
Methylene chloride	CH <sub>2</sub> Cl <sub>2</sub>	+	+	+	+	+	+	+	-	+	-	+	+	+	+	Anhydrous and acid-free
Milk		-	-	-	-	-	-	O	+	-	-	+	+	O	-	-
Molasses		O	O	O	O	O	O	O	O	O	-	+	+	O	-	-
Monochloracetic acid(anhydrous)	CH <sub>2</sub> ClCOOH															
Cold												O	O			Si-cast iron, Hastelloy C
70 °C												-	-			Hastelloy C
Monoethanol amine (MEA)	H <sub>2</sub> N-CH <sub>2</sub> -CH <sub>2</sub> OH	+	+	+	+	+	+	-	+	+	+	+	+	+	-	For reasons of purity 1.4408/9.4460
Naphthalene	C <sub>10</sub> H <sub>8</sub>	O	O	+	+	+	+	+	-	+	+	+	+	+	+	-
Nick chloride, cold	NiCl <sub>2</sub>															
<20%		-	-	-	-	-	-	-	O	O	-	+	+	-		
>20%		-	-	-	-	-	-	-	-	O	-	+	+	-		Hard rubber
Nickel sulfate Cold	NiSO <sub>4</sub>	-	-	-	-	-	-	-	+	+	+	+	+	+	-	
(pure solution 80 °C		-	-	-	-	-	-	-	-	+	-	+	+	-		
Nickel sulfate, aqueous solution	NiSO <sub>4</sub>	-	-	-	-	-	-	-	O	O	O	O	O	O	-	-
Nitrating acid	H <sub>2</sub> SO <sub>4</sub> /HNO <sub>3</sub>	-	-	-	-	-	-	-	O	O	O	O	O	O	-	-
Nitric acid, technical		-	-	-	-	-	-	-	O	O	O	O	O	O	-	It is indispensable to contact ASK!

دول مقاومت شیمیایی مواد

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#### ۰: قابل استفاده در شرایط خاص

- : ضعيف

خالی : آزمایش نشده

Pumped medium														Special materials / remark	
		GG-25	GGG-40;GGG-40.3	GS-C 25	ERN	NI-Resist D2	NORIHARD® NH 15.3	NORLLOY® NL 25.2	1.4408	1.4408	1.4500	NORICID® 9.4306	NORDUR® NC 24.6	G-CuAl10 Ni	
Nitric acid, pure	HNO <sub>3</sub>	-	-	-	-	-	-	-	-	-	-	O	O	O	
Concentration <60%, Cold boiled		-	-	-	-	-	-	-	-	-	-	O	O	-	
Nitric acid, pure Concentration <90%, Cold 98%, boiled Cold boiled	HNO <sub>3</sub>	-	-	-	-	-	-	-	-	-	-	O	O	O	
Nitrobenzene	C <sub>6</sub> H <sub>5</sub> NO <sub>2</sub>	+	+	+	+	-	-	-	-	-	-	O	-	-	
Nitrocellulose		+	+	+	+	-	-	-	-	-	-	O	O	O	
Nitrophenol	C <sub>6</sub> H <sub>5</sub> (OH)NO <sub>2</sub>	+	+	+	+	-	-	-	-	-	-	O	O	O	
Oils		+	+	+	+	-	-	-	-	-	-	O	-	-	
Oleum (20% free SO <sub>3</sub> )		-	-	-	-	-	-	-	-	-	-	O	-	-	
Oxalic acid	(COOH) <sub>2</sub>	-	-	-	-	-	-	-	-	-	-	O	O	O	
Paraffins		+	+	+	+	-	-	-	-	-	-	O	O	O	
Pentanol	C <sub>5</sub> H <sub>11</sub> OH	+	+	+	+	-	-	-	-	-	-	O	-	-	
Petrol(gasoline)		+	+	+	+	-	-	-	-	-	-	O	-	-	
Petroleum (crude oil)		O	O	O	O	-	-	-	-	-	-	O	-	O	
Phenol,aqueous solution	C <sub>6</sub> H <sub>5</sub> OH	-	-	-	-	-	-	-	-	-	-	O	-	-	
Phenolsulfonic acid	HOC <sub>6</sub> H <sub>4</sub> SO <sub>3</sub> H	-	-	-	-	-	-	-	-	-	-	O	-	-	
Phosphoric acid, pure Concentration <85%, Cold 100°C	H <sub>3</sub> PO <sub>4</sub>	-	-	-	-	-	-	-	-	-	-	O	-	-	
Phosphoric acid, technical (Wet hydrolysis) Concentration<75% , Cold 80°C		-	-	-	-	-	-	-	-	-	-	O	O	O	
Phthalic anhydride	C <sub>8</sub> H <sub>4</sub> .(CO) <sub>2</sub> O	O	O	O	O	-	-	-	-	-	-	O	-	-	
Plant oils (vegetable fat)		+	+	+	+	-	-	-	-	-	-	O	-	-	
Potassium aluminum sulfat	KAl(SO <sub>4</sub> ) <sub>2</sub>	-	-	-	-	-	-	-	-	-	-	O	-	-	
Potassium carbonate(potash)	K <sub>2</sub> CO <sub>3</sub>	+	+	+	+	-	-	-	-	-	-	O	-	-	
Potassium chloride	KCl	-	-	-	-	-	-	-	-	-	-	O	-	-	
Potassium chromate	K <sub>2</sub> CrO <sub>4</sub>	O	O	O	O	-	-	-	-	-	-	O	-	-	
Potassium chromium sulphate	KCr(SO <sub>4</sub> ) <sub>2</sub>	-	-	-	-	-	-	-	-	-	-	O	-	-	
Potassium cyanide	KCN	O	O	O	O	-	-	-	-	-	-	O	-	-	
Potassium hydroxide Concentration 30%, Cold 80°C 50%, Cold 80°C	KOH														
Potassium hypochlorite	KOCl	-	-	-	-	-	-	-	-	-	-	O	O	-	
Potassium nitrate	KNO <sub>3</sub>	-	-	-	-	-	-	-	-	-	-	O	O	-	
Potassium oxalate	(COOK) <sub>2</sub>	-	-	-	-	-	-	-	-	-	-	O	-	-	
Potassium permanganate	KMnO <sub>4</sub>	O	O	O	O	-	-	-	-	-	-	O	-	-	
Potassium silicate	K <sub>2</sub> SiO <sub>3</sub>	O	O	O	O	-	-	-	-	-	-	O	-	-	
Potassium sulfate	K <sub>2</sub> SO <sub>4</sub>	-	-	-	-	-	-	-	-	-	-	O	-	-	
Propionic acid, aqueous solution	C <sub>3</sub> H <sub>5</sub> .COOH	-	-	-	-	-	-	-	-	-	-	O	-	-	
Pyridin	C <sub>5</sub> H <sub>5</sub> N	+	-	+	+	-	-	-	-	-	-	O	-	-	
Pyrollidon	HN.(CH <sub>2</sub> ) <sub>3</sub> .CO	+	-	+	-	-	-	-	-	-	-	O	-	-	
Saccharine juice		O	O	O	O	O	O	-	-	-	-	O	-	-	
Seawater															
Selexol		O	O	O								O	-	-	
Silicon oil		+	+	+	+	+	+	+	+	+	+	O	-	+	
Silver nitrate, aqueous solution	AgNO <sub>3</sub>	-	-	-	-	-	-	-	-	-	-	O	-	-	
Slop (distillation residue)		O	O	O	O	-	-	-	-	-	-	O	-	O	
Sodium acetate, aqueous solution	Na.CH <sub>3</sub> COO	-	-	-	-	-	-	-	-	-	-	O	-	O	
Sodium aluminate. aqueous solution concentration <20%,Na <sub>2</sub> O cold 80°C >31,5%Na <sub>2</sub> O cold 80°C	Na <sub>3</sub> [Al(OH) <sub>6</sub> ]														
Sodium carbonate. aqueous solution	Na <sub>2</sub> CO <sub>3</sub>	+	+	+	-	-	-	-	-	-	-	O	-	-	
Sodium chloride, aqueous solution	NaCl	-	-	-	-	-	-	-	-	-	-	O	-	O	
Sodium hydrogen sulfate	NaHSO <sub>4</sub>	-	-	-	-	-	-	-	-	-	-	O	-	O	
Sodium hydrogen sulfate, cold	NaHSO <sub>3</sub>	-	-	-	-	-	-	-	-	-	-	O	-	O	
Sodium hydroxide. aqueous solution Concentration 30%. Cold 80 °C 50%. Cold 80 °C	NaOH														
Sodium hypochlorite Concentration ≤ 80g/l. Cold boiled	NaOCl														
Sodium nitrate, cold	NaNO <sub>3</sub>	+	+	+	+	-	-	-	-	-	-	O	O	O	
Sodium perborate	NaBO <sub>2</sub> .H <sub>2</sub> O <sub>2</sub>	-	-	-	-	-	-	-	-	-	-	O	-	-	
Sodium silicate	Na <sub>2</sub> O <sub>x</sub> SiO <sub>2</sub>	O	O	O	O	-	-	-	-	-	-	O	-	O	
Sodium sulfate, aqueous solution	Na <sub>2</sub> SO <sub>4</sub>	-	-	-	O	-	-	-	-	-	-	O	-	O	
Sodium sulfide, aqueous solution	Na <sub>2</sub> S	-	-	-	-	-	-	-	-	-	-	O	-	O	
Sodium sulfite, aqueous solution	Na <sub>2</sub> SO <sub>3</sub>	-	-	-	-	-	-	-	-	-	-	O	-	O	
Sodium thiosulfate	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub>	-	-	-	-	-	-	-	-	-	-	O	-	O	
Spin bath acid (with H <sub>2</sub> S and CS <sub>2</sub> )		-	-	-	-	-	-	-	-	-	-	-	-		
Stretford base		O	O	O								O	-	O	

Special materials / remark											
		GG-25	GGG-40; GGG-40.3	GS-C-25	ERN	Ni-Resist D2	NORTHARD® NH 15.3	NORLOY® NL 25.2	1.4008	1.4308	NORCICID® 9.4306
Stretford base		O	O	O					+	+	+
Styrene	C <sub>6</sub> H <sub>5</sub> -CH=CH <sub>2</sub>	O	O	+	+				+	+	+
Sulfur, fused	S	+	+	+	+				+	+	+
Sulfur dioxide, aqueous solution	H <sub>2</sub> SO <sub>3</sub>	-	-	-	-			-	+	+	+
Sulfuric acid, pure	H <sub>2</sub> SO <sub>4</sub>										
Concentration	10% Cold	-	-	-				-	-	+	O
	90 °C	-	-	-				-	-	+	+
	40% Cold	-	-	-				-	-	+	+
	40 °C	-	-	-				-	-	+	+
	55-80% Cold	-	-	-				-	-	+	+
	90% Cold	+	+	-				-	+	+	+
	40 °C	-	-	-				-	O	+	+
	98% Cold	+	+	O				-	+	+	+
	70 °C	-	-	-				-	-	+	+
Sulfuric acid , technical		-	-	-				-	O	O	O
Sulfurous acid	H <sub>2</sub> SO <sub>3</sub>	-	-	-				-	+	+	+
Tannic acid	C <sub>76</sub> H <sub>52</sub> O <sub>46</sub>	-	-	-				O	O	+	+
Tar, tar oils		O	O	+	O	+	O	O	+	+	+
Tartaric acid, aqueous solution	(CHOH) <sub>2</sub> .(COOH) <sub>2</sub>	-	-	-	-	-	O	O	+	+	O
Terephthalic acid	C <sub>8</sub> H <sub>6</sub> (COOH) <sub>2</sub>	-	-	-			O	O	+	+	O
Titanium dioxide suspension	TiO <sub>2</sub>	-	-	-	-		-	-	-	+	+
Toluene (methibenzene)	C <sub>6</sub> H <sub>5</sub> -CH <sub>3</sub>	O	O	+	+			+	+	+	+
Trichloroacetic acid	CCl <sub>3</sub> -COOH	-	-	-	-			-	-	-	-
Trichloroethene	CHCl=CCl <sub>2</sub>	+	+	+	+			+	+	+	+
Trisodium phosphate	Na <sub>3</sub> PO <sub>4</sub>	O	O	O	O	+	O	+	+	+	O
Urea (carbamide)	CO(NH <sub>2</sub> ) <sub>2</sub>	O	O	O	+		O	+	+	+	-
Water:	H <sub>2</sub> O										
Boiler feed water		O	O	+	+			+	+	+	O
Brackish water		-	-	-	+	-		+	+	-	O
Brackish water with sand		-	-	-	-	-		-	-	O	O
Chemistry waster water		O	O	O	O	O	O	O	O	O	O
Complete softening of water		-	-	-	-	-	+	+	+	+	+
Condensate		O	O	+	+			+	+	+	O
Condensate (not boiler)		-	-	-	-			+	+	+	+
Cooling water		-	-	-	-			+	+	+	+
Deminaliz. Water =deionat		O	O	O	O	+	O	O	+	O	O
Distilled water		-	-	-	-			+	+	+	+
Drinking water		-	-	-	-			+	+	+	+
Fire water		+	+	+	+	+	+	+	+	+	+
Heating water		O	O	+	+		O	O	+	O	+
Partial softening of water		-	-	-	-		O	O	+	O	+
Pure water		+	+	+	+	+	+	+	+	+	+
Seawater		-	-	-	O	-	-	O	O	-	O
Seawater with sand		-	-	-	-	+	-	-	-	O	O
Sewage municipal		+	+	+	+	+	+	+	+	+	+
industrial		-	-	-	-			+	+	+	+
Sintering water		O	O	O	O	O	O	O	O	O	O
Untreated water		-	-	O	-	+	+	O	+	+	O
Wine		-	-	-	-	-	O	+	+	+	-
Zinc chloride	ZnCl <sub>2</sub>										
Concentration	20%, Cold	-	-	-	O	O	O	+	O	+	-
	80°C	-	-	-	O	-	O	O	-	+	-
	60%, Cold	-	-	-	O	-	O	O	O	+	-
	80°C	-	-	-	-	-	O	O	-	+	-
Zinc sulfate	ZnSO <sub>4</sub>	-	-	-	O	-	O	O	O	+	O
											H <sub>2</sub> SO <sub>4</sub> -free

Analysis of pumped medium  
Analysis of pumped med. is necessary, cor-chem. neutral

Corrosion-chemically neutral

Corrosion-chemically neutral

Conditioned

Depends very much on sand content

Analysis of pumped medium

Conditioned (see guidelines)

Analysis of pumped medium

Conditioned

Inflammable

With specification of analysis

Hard rubber

Hard rubber , Si-cast iron

Anhydrous and acid-free

Hastelloy C

With specification of analysis

It is indispensable to contact ASK!

<130°C

Inflammable

heed isocorrosion curves!